

## **SUPPORTING INFORMATION**

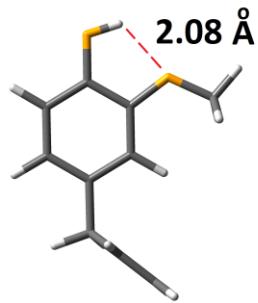
# **DFT Study of the Antiradical Properties of Some Aromatic Compounds Derived from Antioxidant Essential Oils: C-H Bond vs O-H Bond**

Houssem BOULEBD

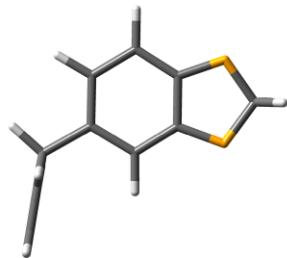
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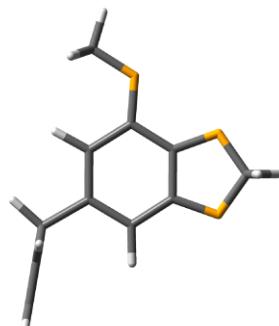
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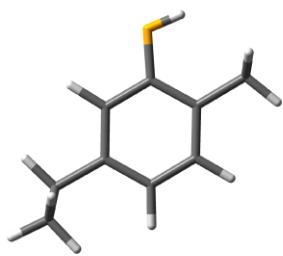
Eugenol (EUG)



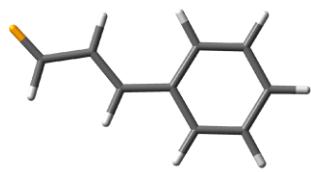
Safrole (SAF)



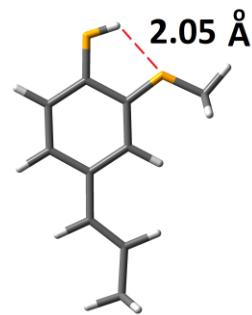
Myristicin (MYR)



Carvacrol (CAR)



Cinnamaldehyde  
(CIN)



Isoeugenol (ISO)

**Fig. S1.** Optimized molecular geometries with intramolecular hydrogen-bonds of EUG, SAF, MYR, CAR, CIN and ISO obtained at B3LYP/6-311G(d,p) level of theory.

**Table S1.** BDE values in kcal/mol of the potential hydrogens of **EUG**, **SAF**, **MYR**, **CAR**, **CIN** and **ISO** calculated at B3LYP/6-311G(d,p) level of theory.

Compound	BDE in kcal/mol		
	Gas	Water	Benzene
<b>EUG (OH)</b>	81.07	78.72	77.96
<b>EUG (CH)</b>	72.47	72.56	69.80
<b>SAF (CH)</b>	72.46	72.61	70.77
<b>CIN (CH)</b>	87.40	88.90	86.02
<b>MYR (CH)</b>	72.20	72.48	70.55
<b>CAR (OH)</b>	79.47	79.65	77.73
<b>CAR (C1H)</b>	86.66	87.38	85.22
<b>CAR (C2H)</b>	82.72	83.05	81.11
<b>ISO (CH)</b>	78.76	79.14	77.17
<b>ISO (OH)</b>	73.85	73.06	71.73

**Table S2.** Calculated Gibbs free energies ( $\Delta G$  kcal/mol) of the reaction of **EUG**, **SAF**, **MYR**, **CAR**, **CIN** and **ISO** with  $\text{HO}^\cdot$  and  $\text{HOO}^\cdot$  at 298.15 K

Compound	Gibbs free energies kcal/mol ( $\text{HOO}^\cdot$ )			Gibbs free energies kcal/mol ( $\text{HO}^\cdot$ )		
	Gas	Water	Benzene	Gas-phase	Water	Benzene
<b>EUG (OH)</b>	-8.33	-4.42	-1.90	-41.05	-37.36	-34.53
<b>EUG (CH)</b>	-8.24	-9.30	-8.82	-40.97	-42.24	-41.45
<b>SAF (CH)</b>	-8.28	-9.56	-8.87	-41.00	-42.50	-41.50
<b>CIN (CH)</b>	8.70	5.97	6.92	-24.03	-26.97	-25.72
<b>MYR (CH)</b>	-8.56	-9.53	-9.02	-41.28	-42.46	-41.65
<b>CAR (OH)</b>	-2.13	-3.06	-2.72	-34.85	-35.99	-35.35
<b>CAR (C1H)</b>	5.24	5.16	5.17	-27.49	-27.78	-27.46
<b>CAR (C2H)</b>	0.25	-0.73	-0.13	-32.48	-33.66	-32.77
<b>ISO (CH)</b>	-1.57	-2.16	-1.83	-34.30	-35.09	-34.47
<b>ISO (OH)</b>	-6.90	-8.83	-7.83	-39.62	-41.77	-40.46